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(7.1) Ions Key Term. To find the number of valence electrons in an atom of a representative element, simply look at its group number.

(7.1) Ions Key Term cont. A positively charged ion, or cation, is produced when an atom loses one or more valence electrons. (7.1)

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STUDY. PLAY. valence electrons.
electrons in the highest occupied

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energy level of an element's atoms. Group Number, 1A= 1, 2A=2, 3A=3, 4A=4, 5A=5, 6A=6, 7A=7, 8A=8. How can you determine the number of valence electrons an atom has from its position on the periodic table? Name by that

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STUDY. PLAY. Anion. An ion with a negative charge. The charge for an Anion is written as a number followed by a minus sign. Cation. An ion with a positive charge. The charge for a cation is written as a number followed by a plus sign.
Chemical formula.

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what follows an element's symbol
in a chemical formula to indicate
the number of atoms of that
element in one molecule of the
compound? what does the chemical
formula for an ionic compound
represent?

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STUDY. PLAY. Explain why most
carbon compounds are classified
as organic compounds. Carbon
compounds controls what happens
in the cell of living organism and
play key roles in the chemistry of
living organisms. Why is lead no
longer used in paints for plumbing
pipes.

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Answers

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Define chemical bond. Explain formation of cations and anions.

Draw electron dot diagrams. Write electron configurations for ions.

The force that holds two atoms together. Cations form when an atom loses electrons. An anion

forms when.... Ex. K- $\rightarrow 1s^2 2s^2 2p^6 3s^2 3p^6 4s^1$.

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Vocabulary. Valence electron.
Electron dot structure. Octet rule.
Halide ion. the electrons in the
highest occupied energy level of
an eleme.... diagrams that show
valence electrons in the atoms of
an elemen.... in forming
compounds, atoms tend to achieve
the electron confi....

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CHEM 109 Chapter 7: Energy
Storage either decreasing the
hydrogen pressure or increasing
the temperature, the H₂ can be re-
released on demand. Metal-hydride
storage systems are ideally suited
for PEM fuel cells that require high-
purity hydrogen. Because metal
hydrides are selective and absorb
only hydrogen and not larger gas
molecules, they act simultaneously
...

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the force that holds two atoms
together; may form by the
attraction of a positive ion for a
negative ion or by sharing
electrons. a three-dimensional

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geometric arrangement of particles in which each positive ion is surrounded by negative ions and each negative ion is surrounded by positive ions; vary in shape due to sizes and relative numbers of the ions bonded.

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STUDY GUIDE FOR CHAPTERS 6 & 7. Certain topics like the structure, synthesis, and nomenclature of alkyl halides have already been covered previously. This part of the course is primarily concerned with S_N1 and S_N2 reactions, $E1$ and $E2$ reactions, and alkene synthesis, structure, and nomenclature. The following is the suggested order of readings and practice problems.

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