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Answers

Chemquest 33

Limiting Reactants

Answers the "limiting
reactant" and oxygen
is the excess

reactant. For each

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mole of C_3H_8 five
moles of O_2 are
required, so for 12.5
moles of C_3H_8 , the
number of moles of O_2
needed are
 $(12.5)(5) = 62.5$
moles.

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Limiting Reactants

Answers the “limiting reactant” and oxygen is the excess

reactant. For each mole of C_3H_8 five moles of O_2 are required, so for 12.5 moles of C_3H_8 , the number of moles of O_2 needed are

$$(12.5)(5) = 62.5 \text{ moles.}$$

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Limiting Reactants

Answers the "limiting reactant" and oxygen is the excess

reactant. For each

mole of C_3H_8 five

moles of O_2 are

required, so for 12.5

moles of C_3H_8 , the

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number of moles of O
2 needed are
 $(12.5)(5) = 62.5$
moles.

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Answers (Base the
answer to this
question on the
number of moles of

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33 propane that actually get combusted—which is your answer to part a.) 12 moles. For every mole of propane that combusts 3 moles of CO₂ are produced, so the number of moles of CO₂ that can be produced when 4 moles of propane combusts = $4(3) = 12$.

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Limiting Reactants

Answers the "limiting
reactant" and oxygen
is the excess

reactant. For each

mole of C_3H_8 five

moles of O_2 are

required, so for 12.5

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33 moles of C_3H_8 , the number of moles of O_2 needed are
 $(12.5)(5) = 62.5$
moles.

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c) $3.1 \times 10^{-34} + 2.2 \times 10^{-33} =$ first, change exponents so both are either -34 or

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33 Answer sheet. Any substance listed on the periodic table. For example, phosphorus or nitrogen.

Chemquest 33
Answer Key -
exams2020.com
ChemQuest 33 Name
T tl't'cym ,uTg.eTInT
F.uT'r \$\$ Date Hour
Information:

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Determining if a Bond is Polar In general the greater the difference in electronegativity between two bonding atoms, the greater the polarity of the bond. A general rule of thumb is that if the difference in electronegativity is less than 0.5 then the bond is considered nonpolar.

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,uTg.eTlnT tlt'cym

F.uT'r - LTHS

Answers

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Name: _____

_____.

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_____.

Hour: _____.

Information: Limiting

Reactant. Again

consider the

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33 Answers

combustion of
propane: $C_3H_8 + 5$
 $O_2 \rightarrow 3 CO_2 + 4 H_2O$.

If you had 10 moles of propane to burn, you would need 50 moles of oxygen according to the ratio in the balanced equation.

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Webs

ChemQuest 10

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Name: _____ Date:

_____ Hour: _____

Information: Bohr's
Solar System Model
of the Atom All the
planets are attracted
to the sun by gravity.
The reason that the
Earth doesn't just
float out into space is
because it is
constantly attracted to
the sun. Similarly, the
moon is attracted to

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Questions: Bohr's
reasoning
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moles of O₂ are
required, so for 12.5
moles of C₃H₈, the
number of moles of O₂
needed are
 $(12.5)(5) = 62.5$
moles. Since we have
more than 62.5 moles
(according to the
question we have

Page 4/26

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Limiting Reactants

Answers ChemQuest

33 Name T tl'cym

,uTg.eTlnT F.uT'r \$\$

Date Hour

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Limiting Reactants

Answers

PLEASE NOTE: If
you have a question
about these answers,

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33 Answers
it is your responsibility
to come to office
hours or ask during
class work time. UNIT
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Keys - ChemActivity
51: Cell Voltage -
ChemActivity 50:
Electrochemical Cell
UNIT 11 - HW
Practice Keys -
ChemQuest 55: Free
Energy - ChemQuest
54: 2nd Law of

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HW Keys - Roosevelt
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Chemistry 2017-18
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Strategy Consulting |

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Name: Date Hour::F-F

Fluorine Gas Molar

mass = 38.0 g c.t- QI

Chlorine Gas Molar

mass ::Br BI Bromine

Liquid Molar mass : I-

I' Iodine Solid Molar

mass : Information:

Some Molecules and

Their States Critical

Thinking Questions

What type of force

exists between two Fz

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38 Answers
molecules--{ispersion,
dipolar, or hydrogen
bonds?,".eh"s#n
?(€l'5Feq-ece\$'1
What t5.pe of force
exists between ...

Critical Thinking
Questions - LTHS
Answers

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needed are
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more than 62.5 moles
(according to the
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_____ Date: _____

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Information: Changes in Matter Books are made of matter. You are made of matter. “Matter” is a fancy word for the “stuff” of which all objects are made. Every day, matter is changed in different ways. For example, paper can be changed in many ways—it can be torn, folded, or burned. ...

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Information : Changes
in Matter - Science! -
Home

A certain element
exists as three
different isotopes.
24.1% of all isotopes
have a mass of 75.23
amu, 48.7% have a
mass of 74.61 amu,
and 27.2% have a
mass of 75.20 amu.

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Chemistry

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LcW s Structures tr I

Name Date: Hour:

Introduction

Questions Draw the

Lewis structures for

the following

molecules. Part c is

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done for you. a) H₂O
(oxygen is central) b)
SO₂ c) N₂ It t-t (a [Jrt
@ N=N @ @ f\ @ 2
In question 1 above
you should see three
different kinds
of bonds of bonds as
best you can.

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7b9ecd670