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**Solutions and**

**Precipitation**

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Note: Not all of the reactions will occur. For those that do not, write no reaction. 1. A piece of copper is dropped into a container of water. No reaction 2.

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Liquid bromine is added to a container of sodium iodide crystals.  $\text{Br (l)}$   
 $+ 2\text{NaI (s)}$   
 $2\text{NaBr (s)} + \text{I}_2 \text{(s)}$

3. An aluminum strip is immersed in a solution of silver nitrate.

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tin(II)

carbonate sodium

hydrogen

carbonate NaHCO



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1. 2. 3. 4. 5.  
6. 8. 9. 10. 11.  
vanadium (V)  
oxide H<sub>2</sub>O  
dihydrogen  
monoxide (NH) CO  
ammonium oxalate  
4 224  
polonium (VI)  
thiocyanate  
Po (SCN) 6  
tetraphosphorus  
decaoxide P O 4  
10 12. 13. 15.

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16. 17. 18. 19.  
manganese (VII)  
oxide copper (II)  
dihydrogen  
phosphate

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balance the  
following single  
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reactions.

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the following  
names:

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all, one must be  
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these eight  
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answer (Note:  
Additional  
responses will  
not be graded or  
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occur. For those  
that do not,  
write no  
reaction. 1.

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